SYNTHESIS OF ANTIBIOTIC FORTIMICIN B

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Aminocyclitol antibiotic fortimicin B has been synthesized by condensation of 2,6-bis-N-(2,4-dinitrophenyl)- α -6-epi-purpurosaminyl chloride and 1,2:4,5-di-N,0-carbonylfortamine B, followed by removal of all protective groups.

Antibiotic fortimicin A ($\underline{1}$) and B ($\underline{2}$) are produced in a fermentaion broth of *Micromonospora olivoasterospora*.^{1,2)} The structures of $\underline{1}$ and $\underline{2}$ have been established by Egan and his coworkers³⁾ by spectroscopic studies combined with chemical degradations. Both antibiotics are unique pseudodisaccharides comprising a 2,6-diaminoheptose derivative named 6-epi-purpurosamine B and a novel *chiro*-inosadiamine-1,4 derivative designated as fortamine B.³⁾ The former sugar component was synthesized in our laboratory, ^{4,5)} and the latter aminocyclitol has been prepared very recently by Sano, Shirahata and Mori.⁶⁾

We have attempted to synthesize fortimicins by coupling a purpurosamine derivative and a fortamine derivative. Now, we wish to report a synthesis of $\underline{2}$, the most important key compound for a preparation of $\underline{1}$. Namely, $\underline{1}$ is readily prepared from $\underline{2}$ by introducing a glycyl group into the methylamino group by a known procedure. 7)

Hydrolysis of methyl 2,6-di-N-acetyl-2,3,4,6-tetradeoxy- β - $\underline{\underline{L}}$ -lyxo-heptopyranoside^{4,5)} (3) in dilute HCl gave 6-epi-purpurosamine B dihydrochloride⁸⁾ ($\underline{\underline{4}}$).

Acetylation of $\underline{4}$ with acetic anhydride and boron trifluoride etherate afforded a salt of 1-O-acetyl-6-epi-purpurosamine B ($\underline{5}$). Reaction of $\underline{5}$ with 2,4-dinitrofluorobenzene and triethylamine in methanol gave 1-O-acetyl-2,6-bis-N-(2,4-dinitrophenyl)-6-epi-purpurosamine B ($\underline{6}$), mp 120-122°C, in 52% yield; ¹H NMR(CDCl $_3$): δ 5.65 (d, $J_{1,2}$ =8.7 Hz, axial H-1), 6.35 (d, $J_{1,2}$ =3.3 Hz, equatorial H-1), a ratio of α : β was approximately 1:2.

Halogenation of <u>6</u> with acetyl chloride in dry ether containing dry hydrogen chloride gave 2,6-bis-N-(2,4-dinitrophenyl)- α -6-epi-purpurosaminyl chloride (<u>7</u>), mp 130-132°C, [α] $_{\underline{D}}^{20}$ +130° (c 0.49, acetone), in 82% yield; ¹H NMR(acetone-d₆): δ 1.45 (d, 3H, J=6.6 Hz, 6-CH₃), 6.72 (d, H, J_{1,2}=3.0 Hz, H-1), 8.69 (d, H, J=9.0 Hz, NH), 8.80 (d, H, J=9.9 Hz, NH).

On the other hand, an aglycon was prepared as follows. Starting from di-N-acetylfortamine $B^{9}(\underline{8})$, by successive acid hydrolysis, deionization and acylation, 1,4-bis-N-(benzyloxycarbonyl)fortamine $B^{8}(\underline{9})$ was prepared in 50% yield, mp 135-136°C, $[\alpha]_{\underline{D}}^{24}$ +48.7° (c 1.1, methanol); ¹H NMR(CDCl₃): δ 3.06 (s, 3H, NCH₃), 3.30 (s, 3H, OCH₃), 5.10 (s, 2H, CH₂), 5.17 (s, 2H, CH₂), 6.46 (d, H, J=9.1 Hz, NH), 7.38 (s, 10H, phenyl).

To avoid a formation of a cyclic carbamate between an amino group on C-1 and a hydroxyl group on C-6, two hydroxyl groups on C-5 and C-6 were blocked by a cyclohexylidene group. Reaction of $\underline{9}$ with 1,1-dimethoxycyclohexane in N,N-dimethylformamide (DMF) in the presence of p-toluenesulfonic acid afforded 1,4-bis-N-(benzyloxycarbonyl)-5,6-O-cyclohexylidene-fortamine B ($\underline{10}$), [α] $\underline{0}^{24}$ +26.1 O (e 0.98, methanol); $\frac{1}{1}$ H NMR(CDCl $_3$): δ 3.07 (s, 3H, NCH $_3$), 3.38 (s, 3H, OCH $_3$), 5.10 (s, 2H, CH $_2$), 5.14 (s, 2H, CH $_2$), 5.63 (broad d, H, J=6.8 Hz, NH), 7.38 (s, 10H, phenyl).

Reaction of $\underline{10}$ with sodium hydride in DMF gave 1,2-N,0-carbonyl-4-N-(benzyl-oxycarbonyl)-5,6-0-cyclohexylidene-fortamine B (11)in 91% yield, mp 71° C (dec.),

[α] $_{\underline{D}}^{24}$ -9.1° (c 0.99, methanol); 1 H NMR(CDCl $_{3}$): δ 1.62 (broad m, 10H, cyclohexylidene), 3.14 (s, 3H, NCH $_{3}$), 3.43 (s, 3H, OCH $_{3}$), 5.17 (s, 2H, CH $_{2}$), 5.52 (broad s, H, NH).

Hydrolysis of $\underline{11}$ with 50% aqueous acetic acid, followed by treating with sodium hydride in DMF afforded 1,2:4,5-di-N,0-carbonylfortamine B ($\underline{12}$) in 54% yield, mp 225-228°C, [α] $\underline{16}$ -84.3° (c 1.06, methanol); IR(KBr) 3360, 1772, 1725 cm⁻¹.

Condensation of $\underline{7}$ and $\underline{12}$ in dioxane in the presence of silver trifluoromethanesulfonate gave 1,2:4,5-di-N,O-carbonyl-2',6'-bis-N-(2,4-dinitrophenyl)fortimicin B ($\underline{13}$) in 37% yield, mp 226-228°C, [α] $_{\underline{D}}^{16}$ +43.2° (c 1.01, acetone); 1 H NMR(acetone-d₆): δ 1.44 (d, 3H, J_{6',7'}=6.3 Hz, 6'-CH₃), 2.85 (s, 3H, NCH₃), 3.61 (s, 3H, OCH₃), 4.67 (dd, H, J=6.0 Hz, J=7.5 Hz, H-5), 5.53 (d, H, J_{1',2'}=3.3 Hz, H-1'), 7.20 (broads, H, 1-NH), 8.82 (d, 2H, J=8.1 Hz, 2' and 6'-NH). Anal. Calcd for C₂₉H₃₂N₈O₁₅ (732): C, 47.54; H, 4.40; N, 15.30%. Found: C, 47.50; H, 4.47; N, 14.99%. Compound 13 was identical with an authentic sample prepared from natural fortimicin B.

Treatment of $\underline{13}$ with Amberlite IRA-400(OH⁻) resin, followed by hydrolysis in barium hydroxide solution gave fortimicin B, $\underline{2}$, $[\alpha]_{\underline{D}}^{16}$ +25.3° (c 0.65, water); 1 H NMR(D₂O): δ 1.07 (d, 3H, J=6.6 Hz, 6'-CH₃), 2.40 (s, 3H, NCH₃), 3.46 (s, 3H, OCH₃), 3.98 (dd, H, J_{4,5}=4.5 Hz, J_{5,6}=9.2 Hz, H-5), 5.03 (d, H, J_{1',2'}=3.0 Hz, H-1'). [lit.²) [α]_{\underline{D}} +22.2° (c 0.1, water)].

N-Acetylation of $\underline{2}$ with acetic anhydride in methanol afforded tetra-N-acetyl-fortimicin B, mp 161-163°C, $[\alpha]_{\underline{D}}^{20}$ +91.6° (c 0.6, methanol). IR and 1 H NMR spectra of the product were superimposable on those of an authentic sample prepared from natural fortimicin B. [lit. 10) mp 155-160°C, $[\alpha]_{\underline{D}}^{25}$ +90.6° (c 0.5, water); lit. 3) $[\alpha]_{\underline{D}}$ +92.72° (c 1.0, methanol)].

References and Note

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